

# CAPTURING CARBON DIOXIDE

Capturing carbon dioxide (CO<sub>2</sub>) from a stationary source such as a power plant involves separating it from the flue gas stream to prevent the gas from being released to the atmosphere.

The main sources for CO<sub>2</sub> capture are natural gas processing, industrial processes, electricity generation and, potentially in the future, hydrogen production.

Natural gas production often involves separating naturally occurring CO<sub>2</sub> mixed with the gas in its natural state before the gas can be sold. As Australia's natural gas industry expands this will be an increasing source of emissions that will need to be dealt with.

Other industrial processes where CO<sub>2</sub> capture is applicable include fertiliser and ammonia production, and cement manufacture; however the total quantity of CO<sub>2</sub> produced by these processes is relatively small.

A far larger source of CO<sub>2</sub>, accounting for approximately half of all CO<sub>2</sub> emissions in Australia, is fossil fuel electricity generation from coal, oil or natural gas. The technology for capturing CO<sub>2</sub> from these sources is currently available, and research is underway to make the process more efficient and cost-effective.

There are three categories of CO<sub>2</sub> capture systems that could be used at power stations: post-combustion, pre-combustion and oxy-firing.

In **post-combustion capture** CO<sub>2</sub> is separated from the flue gas after fuel is burnt. This process can be added, or retro-fitted, to existing power stations, either coal or natural gas-fired.

During **pre-combustion capture** the fossil fuel is reacted with steam and oxygen, producing a synthetic gas (syngas) which is made up of mostly carbon monoxide (CO), carbon dioxide and hydrogen (H<sub>2</sub>). An additional reaction with water (known as a water gas shift) can be used to convert the residual carbon monoxide to CO<sub>2</sub> and additional hydrogen. The CO<sub>2</sub> is removed and the hydrogen can then be burned in gas turbines to produce electricity. Such plants exist today.

This process, where the solid fuel is gasified in either an oxygen or air-blown gasifier, can be applied to all fossil fuels. Examples of this process are Integrated Gasification Combined Cycle (IGCC) or Integrated Drying Gasification Combined Cycle (IDGCC) – an Australian-developed technology.

**Oxy-firing combustion** capture involves the combustion of fuel (coal or gas) in pure oxygen or oxygen-enriched air. The process can produce about 75 per cent less flue gas than air-fueled combustion and the exhaust consists of between 80 and 90 per cent CO<sub>2</sub>. The remaining gas is water vapour, which simplifies the CO<sub>2</sub> separation step. An air separation plant is required to produce pure oxygen for the process from air.

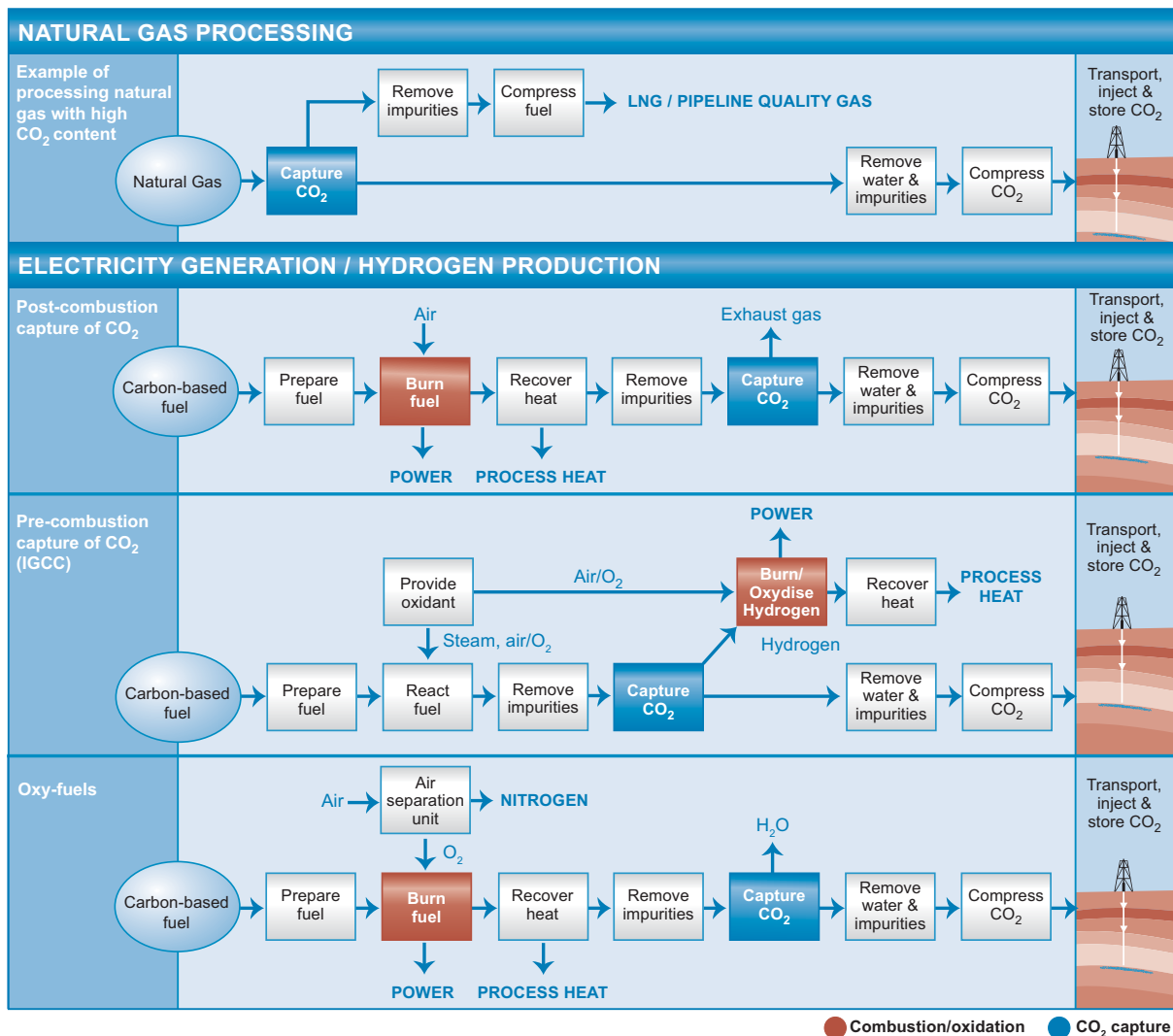


## CO<sub>2</sub> capture has been practised commercially for many years

While the capture of CO<sub>2</sub> for geological storage is a relatively new concept, CO<sub>2</sub> capture for commercial markets has been practised in Australia and overseas for many years.

CO<sub>2</sub> is captured from natural gas wells in South Australia, near Mt Gambier and in southern Victoria, near Port Campbell. The CO<sub>2</sub> is then used for various commercial processes including carbonation of beverages and dry-ice production.

In the United States, CO<sub>2</sub> capture at power plants using chemical absorption based on the monoethanolamine solvent has been practised since the late 1970s, with the captured CO<sub>2</sub> being used for enhanced oil recovery as well as smaller scale CO<sub>2</sub> beverage manufacture.



Processes for carbon dioxide capture